

KOKHNO, N. A.: Master Agric Sci (diss) -- "The forestry properties of maples and their role as stock and a 'second level' in the forest steppe of the Ukrainian SSR". Kiev, 1958. 18 pp. (Min Agric Ukr SSR, Ukr Acad Agric Sci), 150 copies (KL, No 11, 1959, 121)

AUTHOR: Kokhno, N.A. 26-58-2-26/48

TITLE: Changes in the Shape of Leaf Surfaces (Ob izmenenii formy listovoy plastinki)

PERIODICAL: Priroda, 1958, <sup>47</sup> Nr 2, pp 99-100 (USSR)

ABSTRACT: I.V. Michurin has proved that grafting of one species onto another leads to qualitative changes in the plant. In the former Institut lesa Akademii nauk UkrSSR (Institute of Forestry of the Academy of Sciences of Ukrainian SSR) in Goloseyev (Kiyev), P.G. Krotkevich grafted a cutting of Greek walnut on to an oak. The cutting yielded some leaves very different from normal oak leaves.  
The article contains 1 photo.

ASSOCIATION: Botanicheskiy sad Akademii nauk UkrSSR (Botanical Gardens of the Academy of Sciences of Ukrainian SSR, Kiyev)  
Card 1/1

1. Botany--USSR 2. Plants--Grafting 3. Leaves--Characteristics

AUTHOR

Kokhno, N.A.

SOV-26-58-10-37/51

TITLE:

A Rare Case of the Formation of Adventitious Roots by a Maple (Redkiy sluchay obrazovaniya pridatochnykh korney u kleny)

PERIODICAL:

Priroda, 1958,<sup>47</sup> Nr 10, pp 115 - 116 (USSR)

ABSTRACT:

The article deals with the case of the formation of adventitious roots by a maple. The roots formed at the place where a branch had been lopped off. When examined under the microscope, the roots proved to differ considerably from the normal roots, being more similar in structure to normal shoots. The author concludes that they formed as a result of damage to the tissues due to pruning of the branch. There is 1 photo and 3 references, 2 of which are Soviet and 1 American.

ASSOCIATION:

Botanicheskiy sad Akademii nauk USSR, Kiyev (Botanical Gardens of the UkrSSR Academy of Sciences)

1. Trees--Growth

Card 1/1

KOKHNO, M.A.

Natural reproduction of maple in oak forests. Trudy Bot.sada  
AN URSR 6:114-117 '59. (MIRA 13:5)  
(Ukraine--Maple)

KOKHNO, M.A.

Effect of maples on the water balance of oak stands in the  
Ukrainian forest steppe. Visnyk Bot.sada AN UkrSSR no.1:  
120-125 '59.

(MIEA 13:8)

(Ukraine--Plants--Transpiration)  
(Maple)  
(Oak)

17(4), 30(1)

SOV/20-127-2-64/70

AUTHOR: Kokhno, N. A.

TITLE: On the Rate of Root Growth in Acer Platanoides L.

PERIODICAL: Doklady Akademii nauk SSSR, 1959, Vol 127, Nr 2,  
pp 459 - 461 (USSR)

ABSTRACT: No data have hitherto existed on the topic mentioned in the title. The author investigated in 1958 this rate in Acer platanoides L. near Kiyev, in a ten-year-old tree of an oak-maple culture on a grey loamy middle-bluched (podsolic) forest soil. The method of reference 1 was used; it is described as well as data on the plant cover of the site given. The observations showed that the mentioned rate depends in the first place on the humidity of the soil. The temperature of the soil is less important. The small roots grew most quickly during the rain periods. The root decreased somewhat during a hot weather period after the rain. The weakest growth was observed during the dry periods with high soil temperatures, the best growth in June (Table 1). But also the growth of the whole tree was most intensive in June. Numerous root branchings were formed during

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On the Rate of Root Growth in Acer Platanoides L.

SOV/20-127-2-64/70

the dry-hot periods which were almost equal to the growth of the terminal parts within the same interval. Practically it is assumed that the root growth is even with respect to the mass during the vegetation period. This confirms completely the conclusions of reference 2 concerning Acer saccharus Marsh. The author carried out a calculation with respect to orientation and concluded that the growth of the subterranean and superterrestrial part of Acer platanoides L. is equal. By a comparison of the course of the growth within that time an agreement of the highest growth energy of these two tree parts was found. There are 1 table and 2 references.

ASSOCIATION: Botanicheskiy sad Akademii nauk USSR (Botanical Gardens of the Academy of Sciences UkrSSR)

PRESENTED: February 25, 1959, by V. N. Sukachev, Academician

SUBMITTED: February 25, 1959

Card 2/2

KOKHNO, M.A.

Growth rate of the Norway maple and the hedge maple. Vignyk Bot.  
sada AN URSR no. 2:71-74 '60. (MIRA 14:4).  
(Ukraine—Maple)

KOKHNO, M.A. [Kokhno, M.A.]

Biology of mistletoes. Dop. AN URSR no. 11: 1549-1551 '60.

(MIRA 13:11)

1. Botanicheskiy sad AN USSR. Predstavлено akademikom AN USSR P.S. Pogrebnyakom.

(Mistletoe)

KOKHNO, Nikolay Arsen'yevich [Kokhno, M.A.]; POGRENYAK, P.S.  
[Pohrebnjak, P.S.], akademik, otd. red.; KOVAL', V.A.,  
red.; DAKHNO, Yu.M. [Dakhno, Iu.M.], tekhn. red.

[Maple in the wooded steppe regions of the Ukraine; biological  
characteristics and ecology] Kleny lsoostepovych dibrov Ukrainskych  
biologichni osoblyvosti ta ekologija. Kyiv, Vyd-vo Akad. nauk  
URSSR, 1962. 49 p. (MIRA 15:3)

1. Akademiya nauk USSR (for Pogrebnyak).  
(Ukraine--Maple)

"APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000723710002-9

KOKHNO, M.A.

Growing black poplar in the steppe of the Ukrainian S.S.R.  
Visnyk Bot.sada AN URSR no.4:72-75 '62. (MIRA 16:1)  
(Dnepropetrovsk Province—Poplar)

APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000723710002-9"

KONDRATYUK, Ye.M. [Kondratuk, Ie.M.], ovt. red.; ZOSIMOVICH, V.P. [Zosymovych, V.P.], red.; MAKAREVICH, V.A. [Makarevych, V. A.], red.; POPOV, V.P., red.; RUBTSOV, L.I., red.; SOKOLOVSKIY, O.I. [Sokolova'kyi, O.I.], red.; IL'KUN, G.M. [Il'kun, H.M.], red.; KOKHNO, M.A., red.; ANDRIYCHUK, M.D. [Andriichuk, M.D.], red. iad-vy; TUMANOVA, N.A., tekhn. red.

[Biological problems of acclimatized plants] Pytannia biologii aklimatyzovanykh roslyn. Kyiv, 1963. 90 p. (MIRA 16:7)

1. Chlen-korrespondent AN Ukr.SSR (for Zosimovich).  
(Ukraine—Plant introduction)

KOKHNO, N.A. [Kokhno, M.A.]

Ecological conditions of the formation of the ranges of some tree species. Ukr. bot. zhur. 20 no.4:58-64 '63. (MIRA 17:4)

1. Tsentral'nyy respublikanskiy botanicheskiy sad AN UkrSSR.

AUTHORS: Tsukanov, E. F., Ivanchenko, R. K. and Molotkov, L. F.,  
Docents, Pavlenko, B. A., Nikolayev, V. A.,  
Krizhanovskiy, A. L. and Kokhno, P. Ya., Engineers.

Title: Investigation of Loads During Rolling Plates  
(Issledovaniye davleniya pri prokatke listov)

PERIODICAL: 'Stal', 1958, Nr 4, pp 332-334 (USSR)

ABSTRACT: The measurements of rolling loads endured by rolls in a medium plate mill during rolling plates were carried out. The mill consisted of two stands in line: three rolls (IAUP) for rolling plates and two-rolls for riffling plates. In the three roll mill 670 x 517 x 670 mm for rolling smooth plates cast iron rolls with a chilled surface are used and for rifpled plates, forged steel rolls (50 Kg). The length of rolls 1800 mm. In the two roll stand in which only one pass is made for riffling, cast iron rolls of 650 mm diameter with chilled surface are used. The mill is powered with a 900 h.p. motor. Rifpled plate was rolled in 10-12 passes and smooth plates in 11-13 passes. Measurements of loads on rolls were carried out during rolling plates (dimensions in Table 1) and the most characteristic results are given in Table 2. Experimental results are compared in Figs. 1-3. Conclusions: During intensive reductions in cast iron & chilled rolls stresses are formed considerably exceeding the permissible ones. Specified load on rolls

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5-6 kg/mm<sup>2</sup> at the beginning of rolling increases at the end of rolling to 28-30 kg/mm<sup>2</sup>. During rolling on steel rolls the specific load is higher than on rolling on cast iron rolls (due to an increase in friction in the former case). During rolling comparatively thin products (H = 33 mm) the maximum specific pressure was observed at reductions of 34-40%. With further increase in reduction the specific load decreases.

There are 2 tables, 3 figures and 3 references, all of which are Soviet.

ASSOCIATIONS: Dnepodzerzhinskiy vecherniy metallurgicheskiy institut  
(Dneprodzerzhinskiy Evening Metallurgical Institute) and a zavod im.  
Dzerzhinskogo (Works imeni Dzerzhinskogo)

1. Rolling mills--Operation
2. Plates-rolling
3. Rolling mills--Stresses.

KOKHNO, V.I.

Level of initial stresses in high-strength reinforcement wire.  
Stal' 23 no.10:954-955 O '63. (MIRA 16:11)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy  
metallurgii.

KOKHNO, V.I.

Research on the relaxation of wire for prestressed reinforced concrete. Sbor. trud. TSNIICHM no.24, 392-399 '62. (MIRA 15:6)  
(Concrete reinforcement) (Strains and stresses)

KOKHNO, V.I., inzh.

Unit for testing wire reinforcement for relaxation. Bet. i zhel.-bet.  
8 no.3:127-129 Mr '62. (MIRA 15:3)  
(Concrete reinforcement--Testing)

KOKHNO, V.I.

Apparatus for testing wire for relaxation. Zav.lab. 26 no.8:1027-  
1028 '60. (MIRA 13:10)

1. TSentral'nyy nauchno-issledovatel'skiy institut chernoy  
metallurgii.  
(Wire---Testing)

KOKHNO, V.I., inzh.

Effect of straightening on relaxation in wire for prestressed, reinforced concrete. Stal' 21 no. 1:82-84 Ja '61. (MIRA 14:1)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy metalurgii.  
(Reinforcing bars--Cold working)

KOKHNO, V.I.

Testing prestressed-concrete reinforcement wire for creep.  
Zav.lab. 27 no.2:221-222 '61. (MIRA 14:3)

1. TSentral'nyy nauchno-issledovatel'skiy institut chernoy  
metallurgii imeni I.P. Bardina.  
(Wire—Testing)

KOKHNO, V.I.

Testing a growing deformation in vibrating wires used for a  
prestressed reinforced concrete. Zav.lab. 27 no.11:1412 '61.

(MIRA 14:10)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy metallurgii  
imeni Bardina.

(Wire---Testing)

ZOMINNO, V.I.

Effect of initial stresses on the relaxation of wire for stressed  
reinforced concrete. Stal' 22 no.2:186-187 F '62. (MIRA 15:2)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy  
metallurgii.

(Concrete reinforcement)  
(Strains and stresses)

KOKHNO, V.I.

Evaluating the relaxation resistance of wire for prestressed reinforced concrete constructions. Stal' 23 no.2:174-175 F '63. (MIRA 16:2)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy metallurgii.  
(Concrete reinforcement) (Strains and stresses)

KOKHNO, V.I.

Effect of repeated stress and recovery on the relaxation  
resistance of wire for prestressed reinforced concrete. Stal'  
23 [i.e., 24] no.4:375-377 Ap '64. (MIRA 17:8)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy  
metallurgii imeni I.P. Bardina.

MOSHCHINSKAYA, N.K., doktor khim.nauk; POTIYEVSKAYA, S.A.; KOKHNO, Yu.

A.

Water-resistant urea-furyl resin. Bum. i der. prom. no.4:24  
26 O-D '63. (MIRA 17:3)

BOGDANOV, V.A.; KOHNO, Yu.A.; RYABOKLYACH, V.A.

Making furniture by molding thermoplastic synthetic resins. Der.prom.  
8 no.3:13-14 Mr '59. (MIRA 12:4)

1. Ukrainskiy nauchno-issledovatel'skiy institut mekhanicheskoy obrabotki dervva.  
(Plastics--Molding) (Furniture industry)

GITIS, Semen Semenovich; ALEKSEIEV, Vladimir Vasil'yevich [Aleksieiev, V.V.];  
KOKEKO, In.A., otv.red.; TPLYAKOVA, A.S., red.

[Plastics and their uses] Plastichni masy ta ikh zastosuvannia.  
Kyiv, 1960. 38 p. ("Tovarystvo dlia poshyrennia politychnykh i  
naukovykh znan' Ukrains'koi RSR. Ser.7, no.7). (MIRA 13:7)  
(Plastics)

S/123/61/000/02<sup>4</sup>/001/016  
A004/A101

AUTHORS: Mitskevich, Z.A., Shagyan, V.P., Kokhno, Yu.A.

TITLE: The dependence of the physical-mechanical characteristics of caprone parts on the processing methods

PERIODICAL: Referativnyy zhurnal. Mashinostroyeniye, no. 24, 1961, 25, abstract 24A167 ("Tr. n.-i. in-ta mestr. i toplivn. prom-sti", 1960, no. 15, 3 - 24)

TEXT: The authors studied the dependence of the physical-mechanical characteristics of caprone parts on the processing methods and established the possibility of using caprone for food machine parts. Caprone can be used for the manufacture of parts operating in friction units with consistent and liquid lubricants. The friction coefficient and wear of caprone parts on steel with lubrication is considerably lower than the friction coefficient and wear of non-ferrous metals. The load limit on caprone parts without lubrication does not exceed 35 kg/cm<sup>2</sup>. Heat treatment of caprone parts by holding at high temperatures in oil or paraffin ensures a stable crystalline structure which possesses

Card 1/2

The dependence ...

S,123/61/000/024/001/016  
A004/A101

a lower friction coefficient and wear. The authors present the demands on the raw material and the optimum manufacturing conditions of caprone parts by casting.

[Abstracter's note: Complete translation]

Card 2/2

S/081/62/000/004/075/087  
B138/B110

AUTHORS: Mitskevich, Z. A., Shagiyan, V. F., Kokhno, Yu. A.

TITLE: Dependence of the physical and mechanical characteristics of capron components on the method of their production

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 4, 1962, 559; abstract 4P45 (Tr. n.-i. in-ta mestn. i toplivn. prom-sti, no. 15, 1961, 3-24)

TEXT: The influence of methods and conditions of production of capron components on their physical and mechanical properties has been investigated and the possibility of using capron components in food machinery has been studied. Articles were produced by injection moulding on a vertical casting machine (casting pressure of 35-47 kg/cm<sup>2</sup>), on an automatic thermoplastic machine (1500 kg/cm<sup>2</sup>) and from an autoclave (nitrogen pressure of 5 atm). The optimum conditions for converting capron by these methods were deduced. Heat treatment was carried out under infra-red irradiation, and holding at high temperature in oil, paraffin wax and boiling water. A heat treatment consisting in soaking the capron

Card 1/2

SAMSONOV, Vladimir Georgiyevich, inzh.; KHARAKHASH, Viktor  
Georgiyevich, inzh.; MIROSENKO, Nikolay Ivanovich, inzh.;  
SAFONOV, Aleksandr Ivanovich, inzh.; PESIKOV, Ruvim  
Semenovich, inzh.; ALEKSEYEV, Nikolay Nikolayevich, inzh.;  
KOKHNO, Yu.A., inzh., retsentent

[Anticorrosive plastic coatings] Protivokorrozionnye plast-  
massovye pokrytiia. Kiev, Tekhnika, 1965. 89 p.  
(MIRA 18:12)

KAMENETSKIY, Vladimir Yakovlevich; KOKHNO, Yu.A., inzh., retsenzent;  
FURER, P.Ya., red.; GORNOSTAYPOL'SKAYA, M.S., tekhn. red.

[Manufacturing machine and instrument parts from capron] Iz-  
gotovlenie detalei mashin i priborov iz kaprona. Mcskva, Mash-  
giz, 1961. 80 p. (MIRA 15:2)  
(Nylon) (Machinery—Construction) (Instruments)

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8/653/61/000/000/035/051  
I007/I207

AUTHORS: Mitskevich, Z.A., Shagryan, V.F., and Kokhno, Yu.A.

TITLE: Dependence of physicomechanical properties of caprone components on processing methods

SOURCE: Plastmassy v mashinostroyenii i priborostroyenii. Pervaya resp. nauch.-tekhn. konfer. po vopr. prim. plastmass v mashinostr. i priborostr., Kiev, 1959. Kiev, Gostekhizdat, 1961, 376-394

TEXT: Despite the ever-increasing use of polycaprolactame (caprone) in machine building for the manufacture of a great variety of components, the technological processes of their production still need certain improvements in order to obtain components of stable properties. This is a detailed report of experimental investigations on the physicomechanical properties of caprone, and on the study of sanitary properties of ready caprone products. The char -

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8/653/61/000/000/035/051  
I007/I207

Dependence of physicomechanical properties...

acteristics of the starting material are outlined, the effect of processing methods on the properties of the plastics is studied, the influence of machining methods on the properties of the plastics-made components is considered and the wear resistance of caprone is investigated. As was found, caprone may be used for the manufacture of machine elements operating under friction with liquid lubrication provided the maximum (without lubrication) does not exceed 35 kg/cm<sup>2</sup>. The properties of caprone are enhanced by suitable heat treatment (particularly by infra-red radiations). For the use of caprone in the food industry, its low-molecular components have to be preliminarily extracted by methods outlined in this paper. There are 1: figures and 8 tables.

Card 2/2

KOKHNO, Yu.A.

Use of polycaprolactam in the manufacture of machine parts  
for the food industry. Plast.massy no.7:65-67 '62. (MIRA 15:7)  
(Nylon)  
(Food industry—Equipment and supplies)

KOKHNO, Yuriy Arsen'yevich; LEBEDINSKAYA, Emma Abramovna; MEN',  
~~Sof'ya Mikhaylovna~~; SERGIYENKO, Lyudmila Andreyevna;  
FELIKSON, Anna Moiseyevna; SHAGIYAN, Valentina  
Fedorovna; YENIKOLOPOV, N.S., doktor khim. nauk,  
retsenzent

[Polyformaldehyde] Poliformal'degid. Kiev, Tekhnika,  
1964. 90 p.

KOZHNOV, P., inzh., LAPIN, Yu., arkitektor

Lebankment of the Sura River in Pensa. Zhil.-kom. khos. 12  
(MIRA 15:10)  
no. 5:33 My '62.

(Penza - Lebankments)

KOKITIOVER, F.G., inzh.

Precast reinforced concrete retaining walls on pile foundations  
for city embankments. Nov.tekh.zhil.-kom.khoz.: Gor.dor.-most.khoz.  
i transp. no.3:22-29 '63. (MIRA 17:10)

KOKHNOVICH, S.; YAKOVLEVA, I.

Ferric Chloride

Using ferric chloride to precipitate impurities from water. Zhil. -kom. khos.  
2 no. 5, 1952.

9. Monthly List of Russian Accessions, Library of Congress, September 1957, Unclassified  
2

L 4484-66 EWT(m)/FGC/T IJP(c)

ACC NR: AP9024634

SOURCE CODE: UR/0048/65/029/009/1686/1689

AUTHOR: Aglamazov, V.A.; Khazaradze, M.G.; Burduli, A.Y.; Gedevanishvili, L.D.; Kokhodze, L.Sh.; Ponezhiev, M.M.; Sakvelidze, I.I.

ORG: none

TITLE: On fluctuations in the lateral distribution of muons in extensive air showers /Report, All-Union Conference on Cosmic Ray Physics held at Apatity 24-31 August 1964/

SOURCE: AN SSSR. Izvestiya. Seriya fizicheskaya, v. 29, no. 9, 1965, 1686-1689

TOPIC TAGS: secondary cosmic ray, muon, extensive air shower, particle distribution

ABSTRACT: The authors have investigated the lateral distribution of penetrating particles accompanying extensive air showers. The showers were detected by three  $0.07 \text{ m}^2$  scintillation counters at the vertices of an isosceles right triangle having 10 m legs. The position and direction of the shower axis were determined with the aid of six trays of Geiger counters located at the vertices and at the centers of the 70 m sides of an equilateral triangle. The penetrating particle detector was located 200 m.w.e. below the center of the surface assembly and comprised six systems of two  $0.5 \text{ m}^2$  trays of 15 counters each, the two trays of each system being separated by 15 cm of lead. In order to avoid errors due to delta electrons, triggering of two adjacent counters was always ascribed to passage of a single penetrating particle. Data on over  $10^4$  showers with

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ACC NR: AP5024634

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total number of particles ranging from  $5 \times 10^4$  to  $5 \times 10^6$  are tabulated. The lateral distribution of penetrating particles was found to be in good agreement with the formula of S.Bennet and K.Greisen (Physl Rev., 124, 6, 1961). Many more cases were observed in which two or more (up to 5) penetrating particles were recorded in a single shower than can be accounted for by random fluctuations, considering the low flux of penetrating particles and the small area of the detector. It is concluded that correlated groups of muons occur in the column of an extensive air shower, and it is suggested that these may be due to fluctuations in the elementary interaction of ultrahigh energy nucleons. The relation between the frequency of coincidences in the penetrating particle detector and the distance between the two counters involved was in good agreement with that found by L.D.Gedevanishvili and I.I.Sakvarelidze (Soobshcheniya AN GruzSSR, 32, No.2, 297, 1963). In conclusion, the authors express their gratitude to E.L.Andronikavshvili for his guidance of the work, and to M.F.Bibilashvili, R.K.Kazaryan, G.Ye.Chikovani, A.K.Dzhavrishvili, and I.V.Khaldeyev for assistance with the work.

Orig. art. has: 1 formula and 3 figures.

SUB CODE: NP/ S UEM DATE: 00/

ORIG REF: 001/ OTH REF: 001

BC  
Card 2/2

KOKHODEE, T. V.

Dissertation: "Effect of Economic Activity of Human Society on the Change of Fauna in Georgia." Cand Geog Sci, Inst of Geography, Acad Sci USSR, 21 May 54.  
Vechernaya Moskva, Moscow, 12 May 54.

SO: SUM 284, 26 Nov 1954

*BB*

Properties of esteramides of polyimide resins. A. N. Palcheva and  
V. V. Tsvetkov. Zh. polim. i polim. soed., 1965, 28, 257-260.

Properties of polyimide resins in solvents of the type

$\text{R}-\text{CO}-\text{NH}-\text{R}'$ ,  
the formation of the salt compound of the type:  $\text{H}-\text{O}-\text{R}'$ .  
Chain-End-capped polyimides formed through their solvents: most  
likely, take this reaction course. The solvent power is determined  
by the intensity of interaction of the mol. of polyimide with mol.  
of solvent. It is demonstrated by the following series:  
 $\text{HCO}_2\text{N}$  (measured solubility at room temp. both in 50 and 90%  
solid) > 100%  $\text{AcOEt}$  (solubility 6, 40-6, and 100% at 20°, 50°, and  
100°), > phthalic anhydride > higher fatty acids > alcohols  
and  $\text{H}_2\text{O}$  (6.5-45% at 10° in 90%  $\text{MeOH}$ ). The experimental  
measurements of specific viscosity ( $\eta_{sp}$ ) and intrinsic viscosity ( $\eta_{in}$ ) con-  
traction in c.c. per g. of resin dissolved in 100 g. of solvent, and  
(density of resin in solution) were carried out with a resin obtained  
by polymerization of hexamethoxymethylbenzene with and without the content  
of monomer and polymer of lower mol. wt. Because the mol. wt. of  
polyimide resins is comparatively low and because of the distance

*over*

separating polar group (CO-NH) in the chain, the mol. are sufficiently flexible without appreciable solvation and  $\delta$  therefore decreases only slightly with increase of polyamide concn. and does not change appreciably in different solvents. These values of  $\delta$  for 5% solutions of polyamide purified from monomer in 99.5% H<sub>2</sub>O<sub>2</sub> (I), 90% HCOOH (II), and 90% phenol (III) are 1.1768, 1.162, and 1.162, and decreases on going to 10% for the same solvents and same monomers are -0.01, +0.01, and -0.06.  $\delta$  up in I, II, and III for 0.05 and 2% solutions of polymer are 0.096, 0.31, and 0.099 and 0.24, 0.049, and 0.03 and [e] in 0.05% solutions of polymer are 1.01, 0.78, and 0.65. Monomer (benzene) and benzene (CCl<sub>4</sub>) increases the solvated polar groups (CO-NH) and weakens intermol. forces. They therefore draw near of solution in HCOOH. With phenols this does not take place because of the pronounced interaction between phenols and the CO groups in lactam and COCl<sub>2</sub>. J. B. J. ZABA.

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CA

The properties of solutions of polyamide resins. A. N. Pakshver and T. N. Kothiyambar (Leningrad Inst. Chem. Technol.). *J. Applied Chem. U.S.S.R.* 23, 1045 (1970) (Engl. translation), *J. C. S.* 46, 6370. A study was made of the intrinsic  $\eta$ , vol. contraction on mixing a soln. with the dissolved substance, and the nominal d. of a polyamide resin (I) in soln. in 92.5%  $H_2SO_4$ , 90%  $HCO_2H$ , and 90%  $PhOH$ . Three different samples of I, prep'd by polycondensation of caprolactam, were used: (1) contg. no monomer (lactam) or low-mol. admixts. (after extn. with  $H_2O$  at 10°), (2) contg. monomer and low-mol. impurities (not extn. with hot  $H_2O$ ), and (3) contg. no monomer but only low-mol. admixts. Dens. were also made of the percentage solv. of extd. resin (1) in org. acids and alcs. It is concluded that in I the mols. are linked together by powerful intermol. bonds between the CONH groups of neighboring mols. During soln. in compds. of the type ROH, where R = H,  $Me(CH_2)_n$ ,  $Me(CH_2)_nCO$ ,  $SO_2R'$ ,  $NO_2$ , Ph, or  $C_6H_5Me$ , the mobile H atom of these compds.

reacts with the CONH group of the polyamide mols. and breaks the intermol. CO-NH bonds. Solvation of the individual mols. of polyamide takes place. During this process the individual chainlike mols. of polyamide are freed and acquire, most probably, a coiled form. The strength of a solvent is governed by the intensity of reaction of the mols. of polyamides with the mols. of solvent of the ROH type. In respect to solvent strength, different compds. of the ROH type can be arranged in the order:  $HCO_2H > phenols > inorg. acids and H_2O > aliphatic acids > alcs. and H_2O$ . Owing to the relatively low mol. wt. of I and the distance between the neighboring polar CONH groups in the chain, these chain mols. possess adequate elasticity without considerable solvation; for this reason the d. of I falls only slightly with rising concn. of I in soln. and undergoes little change in different solvents. Monomer, e.g., lactam, and ketones, e.g., acetone, solvate the polar CONH group, weaken the intermol. bond, and depress the  $\eta$  of the soln. (in inorg. acid). This phenomenon is not observed in PhOH owing to the strong reaction of phenols with the ketonic group of lactam and acetone. Differences between I and cellulose acetate are pointed out. N. H. J.

PAKSHVER, A.B.; KOKHOMSKAYA, T.N.; DOLININ, R.I.

Influence of mixed solvents on the process of solution of high molecular substances. Zhur. Priklad. Khim. 23, 990-7 '50. (MLRA 3:8)  
(CA 47 no.14:7253 '53)

KOKHOMSKAYA, T. N.

KOKHOMSKAYA, T. N. -- "Investigation of the Toughness of a Polyamide Smelting." Min Higher Education USSR, Moscow Textile Institute, Moscow, 1956. (Dissertation for the Degree of Candidate of Technical Sciences)

SO: Knizhnaya Letopis' No 43, October 1956, Moscow

Kok hoy skagin, Th.

*M*  
The viscosity of poly  
mer A + B (diphenyl carb  
diimide) was  
const. vis. wt. M, the  
methylenedipropamide and  
caprolactam polymer (II)  
the temp. rose from 215  
and B being const. He  
 $\eta = A + M^{\alpha}$ , A and  
B being const.  $\eta$  of II

*He*  
Date: 18.10.92 (1992)  
Viscosity  $\eta$  of polymer V.V.-Reza  
mer (II) was greater than  $\eta$  of  
Both  $\eta$  values decreased  
 $\eta = 280^{\circ} \times \text{low } V + 100$   
 $\eta = 250^{\circ} \times \text{high } V + 100$   
B being const. Low  $\eta$  of polymer

*HE2c*

*2 May*

*Rm day*

KOKHOMSKAYA, V.V.

"Synthesis and Transformation of Alpha-Keto-oxides." Cand Chem Sci, Department  
of Physicomathematical and Technical Sci, Acad Sci Belorussian SSR, Minsk, 1954.  
(KL, No 1, 1955)

Survey of Scientific and Technical Dissertations Defended at USSR Higher  
Educational Institutions (13)  
SO: Sum. No. 598, 29, Jul 55

KOKHOMSKITYH, V.Y.

62-1-12/21

AUTHORS: Nazarov, I. N.; Akhrem, A. A.; Kokhomskaya, V. V.

TITLE: Alpha-Ketooxides. Part 8. Conversions of Alpha-Dioxide of Beta, Beta-Dimethyl Divinyl Ketone (Alfa-Ketookisi. Soobshcheniya 8, Prevrascheniya alfaadiokisi beta, beta-dimetildivinylkotona)

PERIODICAL: Izvestiya Akademii Nauk SSSR, Otdeleniye Khimicheskikh Nauk, 1957, No. 1, pp. 80-90 (U.S.S.R.)

ABSTRACT:  
~~SECRET~~  
This report is devoted to the study of the reaction occurring between alpha-ketodioxide and alcohols, acetic acid, amines and other compounds containing active hydrogen. Hydrolysis of alpha-ketooxide with water at room temperature leads to the formation of 2,2-dimethyl-3,5-dioxytetrahydro-4-pyrone in two stereoisomeric forms - crystalline and liquid. Hydrogenation of 2,2-dimethyl-3,5-dioxytetrahydro-4-pyrone with hydrogen at 120 atm. in the presence of Raney's nickel gives a high yield of 2,2-dimethyl-3,4,5-trioxytetrahydropyran which easily acetylates under the effect of acetic anhydride into a certain



KOKHOMSKAYA, V.V.

AUTHORS: Bardyshev, I. I., Kokhomskaya, V. V. 79-2-57/64

TITLE: Resinic Acids (Smolyanyye kisloty). I. On the Nature of  $\alpha$ -Sapinic Acid (I. O prirode  $\alpha$ -sapinovoy kisloty).

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 2, pp. 538-542 (USSR)

ABSTRACT: The  $\alpha$ -sapinic acid was first isolated from *pinus maritima* Mill by Dupont & Dubourg (ref. 1). The works by V. N. Krestinskiy et al. (ref. 5), Y. M. Akulovich (ref. 6), D. V. Tishchenko et al. (ref. 7) and B. A. Arbuzov (ref. 8) dealt with the first investigations of their structural formulae and properties. It could be concluded that the majority of the authors regard  $\alpha$ -sapinic acid to be a single compound. The present investigations, however, proved that these hypotheses were wrong and that  $\alpha$ -sapinic acid is a mixture of levo-pimamic acid, palustrinic acid, neo-abietic acid, abietic acid, and dextro pimamic acid. The individual compounds were isolated by gradual treatment with bornyl amine, boric acid or diethylamine and fractional crystallization. Ultraviolet absorption spectra were recorded and the various specific data of the individual acids were stated. The palustrinic acid was isolated for the first time. There are 5 figures, 1 table, 13 references, 9 of which are Slavic.

*Inst. Chemistry, AS Belo SSR*

AKHREM, A.A.; KOKHOMSKAYA, V.V.

Heterocyclic analogs of corticosteroids. Part 2: Syntheses  
based on 2,2-dimethyl-4-tetrahydrothiopyrone. Izv. AN SSSR.  
Ser. khim. no.12:2156-2165 D '63. (MIRA 17:1)

I. Institut fiziko-organicheskoy khimii AN BSSR i Institut  
organicheskoy khimii im. N.D. Zelinskogo AN SSSR.

KOZHEVNIKOV, A.R., prof.; POPOVA, G.I., dots.; VOROZHTSOV, I.P.,  
kand. tekhn. nauk, dots.; GERASENKOVS, B.I., kand. sel'-  
khoz. nauk; YUMAGULOV, G.L., kand. sel'khoz. nauk;  
MAR'YASOV, V.G., assistant; VINOGRADOVA, N.I., kand. sel'-  
khoz. nauk; ROKTANEN, L.P., dots., kand. biol. nauk;  
KOKHOMSKIY, F.M., Geroy Sotsialisticheskogo Truda, zasl.  
zootekhnik RSFSR; MAKHNOVSKIY, M.K., dots., kand. ekon.  
nauk; ARTAMONOV, F.D., assistant; MAKAROVA, I.V., red.

[Corn in the Virgin Territory and Western Siberia] Kukuruza  
v tselinnom krae i Zapadnoi Sibiri. Moskva, Kolos, 1965.  
229 p.  
(MIRA 18:9)

1. Omskiy sel'skokhozyaystvennyy institut im. S.M. Kirova  
(for Kozhevnikov, Popova, Mar'yasov, Vinogradova, Kokhomskiy,  
Makhnovskiy, Artamonov). 2. Zamestitel' direktora po nauchnoy  
rabote Severo-Kazakhstanskoy optychnoy stantsii (for Yumagulov).  
3. Zaveduyushchiy laboratoriyye kukuruzy Sibirskogo nauchno-  
issledovatel'skogo instituta sel'skogo khozyaystva (for  
Gerasenkov). 4., TSelinogradskiy sel'skokhozyaystvennyy institut  
(for Roktanen).

KOKHONOV, F.

Higher quality finance work. Fin.SSSR 15 no.11:28-32 N°54.  
(MLRA 8:2)

1. Ministr finansov Belorusskoy SSR.  
(White Russia--Finance)

KOKHONOV, F.  
KOKHONOV, F.

Closer ties of financial organs with the economy. Fin.SSSR 18  
no.9: 73-78-S: '57. (MIRA 10:10)

1. Ministr finansov Beloruskoy SSR.  
(White Russia--Economic conditions) (White Russia--Finance)

KOKHONOV, F.

Everyone out to uncover and mobilize hidden potentialities.  
Fin. SSSR 19 no.2:35-37 F '58. (MIRA 11:3)

1. Ministr finansov RSFSR,  
(White Russia--Finance)

KOKHONOV, P.

On the 40th anniversary of the White Russian S.S.R. Fin,SSSR 19  
no.11:22-28 N '58. (MIRA 12:7)

1. Ministr finansov Belorusskoy SSR.  
(White Russia--Economic conditions)

KOKHONOV, F.

In the struggle to fulfill the resolutions of the 21st Congress.  
of the CPSU. Fin.SSSR 20 no.8:18-26 Ag '59.

(MIRA 12:11)

1. Ministr finansov Belorusskoy SSR.  
(White Russia--Finance)

KOKHONOV, F.

Mobilize hidden potentialities in the economy more fully. Fin.  
SSSR 22 no. 7:34-38 Jl '61.  
(MIRA 14:7)

1. Ministr finansov Belorusskoy SSR.  
(White Russia—Finance)

KOKHONOV, F.

Reveal more fully the hidden potentialities of the economy.  
Fin. SSSR 23 no.3:31-35 Mr '62. (MIRA 15:3)

1. Ministr finansov Belorusskoy SSR.  
(White Russia---Finance)

KOKHONOV, F.

Exercise a stronger control over enterprises. Fin. SSSR 23  
no. 12:26-29 D '62. (MIRA 16:1)

1. Ministr finansov Belorusskoy SSR.

{White Russia—Finance)  
(White Russia—Industrial management)

KOKHONOV, Kh.B.

PERIOD 1 FOOD & DRINKS  
SERIAL 5055

*Veseyozhnye avvenchany po stekloobrazuyushchim sostoyaniyam*. M., Leningrad, 1959.

(Series: Ia: Fruct.)

**Sponsoring Agencies:** Institut khimii silitkow Akademii nauk SSSR. Vsesoyuznyye initsiativnyye obshchestva i zemstvo Dn. Mendeleyeva i Gannarova vremennyy otdeleniya opticheskoy i zashchitnoi zemstvo SSSR. Vserossiyskaya.

**Editorial Board:** A.I. Avgustinskii, V.P. Berezovskiy, N.A. Besborodov, O.K. Botvinkin, V.V. Vargin, A.G. Vinogradov, A.N. Vinogradov, A.M. Vlasov, K.S. Zetserberg, A.A. Lutscher, M.A. Mar'yasova, V.S. Molchanov, R.L. Myshler, Ye.I. Pervyi-Kotilina, T. I. Tsvetova, V.I. Yashkin; Ed. of Publishing House: I.Y. Savchenko, Tech. Ed.: V.P. Kochubev.

**PURPOSE:** This book is intended for researchers in the sciences and technology of

Vitreous State (Cont.)  
 Sov/5055  
 Ayzen, A.A., and Iam-Pu-hai. Boric and Aluminoboric Anomalies of Silica Glass 497  
 Galant, Ye. J. Refractive Index and Coordination Transformations of Aluminoborosilicate Glasses 499  
 Endakov, S.P. On the Structural Transformations in Glasses Containing  
 2.5% Boron 502  
 Pal'mov, Iu. I. Thermodynamical Study of Soda Borosilicate Glasses 507  
 Popovitko, I.A. On the Structure of Soda Borosilicate Glass Subjected  
 to Long Heat Treatment 511  
 Makarov, Iu.B. Effect of Heat Treatment on the Non-Temperature Thermal  
 Conductivity of Soda-Borosilicate Glass 514  
 Pervozvannyy, Yud. [Doctor of Physics and Mathematics], S.P. Endakov,  
 and Iu.B. Andreyev. On Some of the Debatable Problems Relating to the  
 Structure and unusual Properties of Soda Borosilicate Glasses 517  
 Card 21/2

Vitreous State (Cont.)  
 Sov/5055  
 Discussion 521  
 First Session of the Conference  
 On the State and on the Properties of Glass Connected With the Solution of Glass  
 Structure Problem (Proceedings of the Third All-Union Conference Held  
 During September 16-21, 1959)  
 AVAILABILITY: Library of Congress

21/200/102  
 6-85-63

Card 22/2

RENSKIY, Nikolay Mikhaylovich; KOKHOV, A.P., retsensent; KHUTIN, G.I.,  
retsensent; KITA, V.P., red.; SHLEMKOVA, Z.V., red.izd-va;  
BODROVA, V.A., tekhn.red.

[Manual for marine mechanics] Pособие судовому мотористу.  
Moskva, Izd-vo "Technoii transport," 1960. 285 p.

(MIRA 13:12)

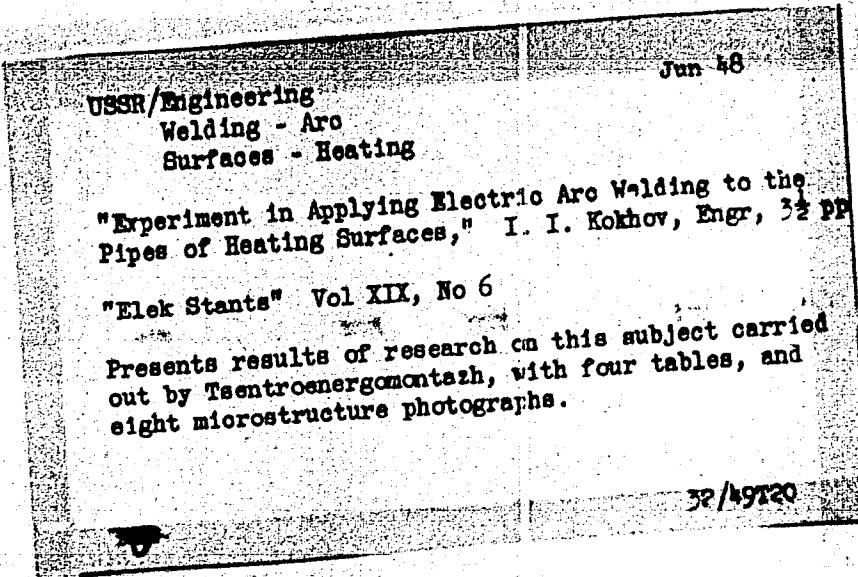
(Marine engines)

PLAKHOV, Veniamin Semenovich; GOGIN, A.F., inzh., retsenzent; KOKHOV,  
A.F., inzh., retsenzent; OSIPOV, I.L., inzh., retsenzent;  
TAREYEV, V.M., prof., doktor tekhn. nauk, red.; VITASHKINA,  
S.A., red. izd-va; BODROVA, V.A., tekhn. red.

[Marine diesel engines; design and operation] Sudovye dizeli;  
konstruktsiya i eksploatatsiya. Moscow, Izd-vo "Rechanoi trans-  
port," 1961. 423 p. (MIRA 15:3)  
(Marine diesel engines)

PA 32/49T20

KOKHOV, I. I.



KOKHOV, M. V. (Dnepropetrovsk)

"The study of the fluctuations of density and concentration with the aid of the dispersion of x rays."

Report presented at the Fourth All-Union Conference on the Liquid State of Matter.

Kiev State Univ., 1-4 June 1961

KOKHOVA, G. M.

AUTHORS: Blok, N. I., Glazova, A. I., Kokhova, G. M. 32-2-6/60  
Lashko, N. F.

TITLE: The Phase Analysis of Complex Titanium Alloys  
(Fazovyy analiz slozhnolegirvannykh titanovykh splavov)

PERIODICAL: Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 2, pp. 141-145  
(USSR)

ABSTRACT: In an earlier work various technical titanium alloys containing aluminium, chromium, molybdenum and changing amounts of hydrogen were already investigated, as was the phase composition of azotized titanium. For the separation of phases a method of the anodic decomposition of alloys was developed. The authors worked with potassium rhodanide, citric acid, glycerin and methanol, at a current density of  $0,013 \text{ A/cm}^2$ , a terminal voltage of 30 V, at from  $-7^\circ$  -  $-10^\circ\text{C}$ . After the electrolysis the anode precipitates were investigated chemically as well as radiographically. In earlier works the Ti-alloys had been smelted in graphite crucibles, the carbon disturbing further investigations; therefore the authors smelted two-to threetimes in arc

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32-2-6/60

## The Phase Analysis of Complex Titanium Alloys

furnaces (till homogenization occurred). The radiograms of heat after-treated (1, 10, 50 hours at 500°C) anode deposits showed the metal stable  $\alpha$ -phase while the  $\omega$ -phase was not observed. The changes in the aging process of the  $\beta$ -phase of two technical alloys (5.08% Al, 3.06% Cr and 4.7% Al, 1.86% Cr, 1.55 % Mo) were put down in a table and the authors noted that after an aging at 450°C only the  $\beta$ -phase is observed while the eutectoid reaction  $\beta - \alpha + Cr_2Ti$  did not take place. Titanium hydride was isolated for the first time and the authors found that hydrogen dissolves mainly in the  $\beta$ -phase (this was found in collaboration with A. T. Yakimova), if, however, there is no such phase the excess hydrogen then forms the titanium hydrides. According to radicgraphic structural analyses the Ti-hydride was of crystalline structure of the NaCl-type, while the neutron-diffraction showed a tetragonal structure. The analyses of the anode precipitates treated in a nitrogen current at high temperatures showed that they consist of one or two phases, the wellknown finely grained TiN and in lower layers the second nitride  $Ti_{n}N$ . The latter is of tetragonal structure. The investigations

Card 2/3

The Phase Analysis of Complex Titanium Alloys

32-2-6/60

of Palty, Margolin and Nielsen concerning the Ti-N system in the  $\zeta$ -phase showed a similar structure, the difference however, between the radiograms found by them and the radiograms of the present work, is considerable. There are 5 tables, and 3 references, 1 of which is Slavic

AVAILABLE: Library of Congress

1. Titanium alloys-Phase studies

Card 3/3

KARAPETYAN, I.S.; KOKHOVA, G.M. (Moskva)

Orthodontic spring lock. Stomatologiya 42 no.4:100-101 Jl-Ag '63  
(MIRA 1714)

1. Nauchno-issledovatel'skiy institut eksperimental'noy khirurgicheskoy apparatury i instrumentov.

KOKHOVA, G. M., Cand Tech Sci (diss) -- "The effect of thermal treatment on the phase transformations and mechanical properties of VT2, VT3, and VT3-1 titanium alloys". Moscow, 1960. 11 pp (All-Union Order of Lenin Sci Res Inst of Aviation Materials), (KL, No 15, 1960, 135)

KOKHOVIT (S. 1)

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S/762/61/000/000/011/029

AUTHORS: Vinogradova, Ye.A., Kokhova, G.M., Lashko, N.F.

TITLE: Phase analysis of heat-treated BT3 (VT3) and BT3-1 (VT3-1) alloys.

SOURCE: Titan v promyshlennosti; sbornik statey. Ed. by S. G. Glazunov. Moscow, 1961, 121-130.

TEXT: The paper comprises a status report on an experimental investigation of the phase composition of the VT3 and VT3-1 alloys of the Ti-Al-Cr-Mo system and the effects thereon of heat treatment. Some of the source data are drawn from Blok, N.I., et al., Zavodskaya laboratoriya, no.2, 1958, 141, and pp.112-120 of the present compendium (Abstract S/762/61/000/000/010/029). At working temperatures the VT3 alloy is found to be two-phase, with an  $\alpha$ - (or  $\alpha'$ -) phase matrix. During cooling of the alloy from high temperatures (HT), the HT  $\beta$  phase may undergo one of three transformations: (1) During fast cooling the  $\beta$  phase transforms into the metastable phase  $\alpha'$ ; (2) during fairly slow cooling the  $\beta$  phase transforms into the metastable  $\alpha'$  phase and some residual  $\beta$  phase; (3) very slow cooling leads to the eutectoid decomposition  $\beta \rightarrow \alpha + \text{TiCr}_2$ . A residual  $\beta$  phase, enriched with Cr, will also form, both during very slow cooling and in the process of aging. In the latter, the alloying elements are redistributed between the  $\alpha$  and the  $\beta$  phase.

Card 1/3

Phase analysis of heat-treated BT3 (VT3)...

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An increase in the aging temperature of VT3 and VT3-1 alloys to 500-550°C enhances the enrichment of the residual  $\beta$  phase with alloying elements. The hardness (H) and ductility (D) characteristics of these alloys can be greatly varied by increases in the aging T after single quenching. The changes in H and D and in the lattice parameters are graphed against T (at 50°C intervals, up to 700-800°). Greatest H and smallest D is obtained upon quenching and subsequent aging at 450-550°. This is attributed to decomposition of the  $\alpha'$  phase, separation of a dispersive  $\alpha$  phase, and - to some extent - the state of the  $\beta$  phase. Thus, the properties of the two alloys after heat treatment are governed by the state, distribution, and form of the particles of  $\alpha$ ,  $\alpha'$ ,  $\beta$ , and TiCr<sub>2</sub> and the redistribution of the alloying elements, and not by the formation of a metastable  $\omega$  phase erroneously postulated by others. The tests comprised melts with differing contents of alloying elements, prepared in vacuum arc furnaces with consumable electrodes, in which Ti and a 50:50 Al-Cr ligature and a 60:20:15 Al-Cr-Mo ligature are fused. The chemical melt composition is tabulated. Phase analysis after 850° quench ( $\alpha+\beta$ -phase region) and 980-990° quench ( $\beta$ -phase region) and 400-700° aging was performed by the powder method (3 full-page tabulations and graph). The total  $\beta$ -phase content did not exceed 8.5% by weight, while the  $\alpha$  (or  $\alpha'$ ) content did not go below 90%. A total  $\beta$ -phase quench was not achieved. The hydrogen content, which could possibly have been responsible for brittleness, was within the bounds specified

Card 2/3

Phase analysis of heat-treated BT3 (VT3) ...

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by the Technical Specs. No Ti hydride was found. The geometric characteristics of the various phases, including the laminar nature of the  $\beta$  phase, the acicular form of the  $\alpha'$  phase, the appearance of an overall basketlike structure, and the segregation of  $\alpha$  phase as a continuous edging at the grain boundaries are described in detail. Desirable avenues for future research are outlined. There are 4 tables and 3 figures; 2 Russian-language Soviet references are cited at the beginning of the text.

ASSOCIATION: None given.

Card 3/3

KOK MOUA, V.F.

**DEHYDRATION PRODUCTS OF THE SYSTEM Na<sub>2</sub>O-Na<sub>2</sub>CO<sub>3</sub>-NaOH-H<sub>2</sub>O AT 25°**

Y. S. Ilina and V. V. Kostin  
 Izdat. Selskogo Prik. Nauk., Akad. Nauk. SSSR, 1957, No. 3, p. 23-28 (1953); cf. C.A. 44, 920; 48, 313d.—The quinary system was studied as 2 quaternary systems: Na<sub>2</sub>SO<sub>4</sub>-Na<sub>2</sub>CO<sub>3</sub>-NaOH-H<sub>2</sub>O and Na<sub>2</sub>SO<sub>4</sub>-NaCl-NaOH-H<sub>2</sub>O. In addition, also was studied the inside space of the quinary system. The NaOH concn. in this investigation was 0-48%. In the system Na<sub>2</sub>SO<sub>4</sub>-NaCl-NaOH-H<sub>2</sub>O the largest crystal area is taken up by Na<sub>2</sub>SO<sub>4</sub>. As the concn. of NaOH was increased from 0 to 48%, the poly. Na<sub>2</sub>SO<sub>4</sub> fell from 31.8% to 0.46%. The presence of NaCl reduced the solv. of Na<sub>2</sub>SO<sub>4</sub> in the presence of NaOH at further. NaCl and NaOH had also a dehydrating effect. Na<sub>2</sub>SO<sub>4</sub>·10H<sub>2</sub>O changed to Na<sub>2</sub>SO<sub>4</sub> in the presence of 14.0% NaCl. A similar change was brought about by 7.7% NaOH. In the system Na<sub>2</sub>SO<sub>4</sub>-Na<sub>2</sub>CO<sub>3</sub>-NaOH-H<sub>2</sub>O 0.67% NaOH induced the formation of 2Na<sub>2</sub>SO<sub>4</sub>·Na<sub>2</sub>CO<sub>3</sub>. At this point the double salt was in equil. with the dehydrates both salts. Higher NaOH concns. only one of the dehydrates was in equil. with the double salt. In the presence of the double salt only 8.0% NaOH was needed for the dehydration of Na<sub>2</sub>SO<sub>4</sub>·10H<sub>2</sub>O. Na<sub>2</sub>CO<sub>3</sub>·aq. was more resistant to the dehydrating effect of NaOH. In the presence of the double salt Na<sub>2</sub>CO<sub>3</sub>·10H<sub>2</sub>O → Na<sub>2</sub>CO<sub>3</sub>·7H<sub>2</sub>O occurred at 9.13% NaOH. The heptahydrate existed at 9.13-12.01% NaOH when it changed to the monohydrate. The latter existed at 12.01-12.1% NaOH when it changed to the anhyd. form. The formation of 2Na<sub>2</sub>SO<sub>4</sub>·Na<sub>2</sub>CO<sub>3</sub> was enhanced by increasing concns. of NaOH up to 50-60% NaOH in the sum NaOH + Na<sub>2</sub>SO<sub>4</sub> + Na<sub>2</sub>CO<sub>3</sub> = 100. Higher concns. of NaOH had no effect. The effects of NaCl and NaOH on dehydration of Na<sub>2</sub>SO<sub>4</sub> and Na<sub>2</sub>CO<sub>3</sub> hydrates, on the formation of the double salt, as on the relative fields of crystal, was studied on inside cut of the quinary system.

M. Hirsch

Inst. Gen. & Organo. Chem. im. N. S. Kur'yanov, AS USSR

Kaklova, V. I.

A study of an equilibrium solid phase  
the system  $K_2CO_3-Na_2CO_3-H_2O$  at temper-  
ature 100°. I. Marzina, L. I. Ilina, and V. I.  
Sokolova. *Fiz.-Khim. Anal., Inst. Obrabotki*  
*Alkali, Nefti S.S.R.*, 25, 307-314 (1954).  
at 100° and 150°, and also x-ray and theographic study  
of  $K_2CO_3-Na_2CO_3$  show that the isotherm of solv. at 100°  
consists of 4 branches corresponding to the  
cryst. of  $Na_2CO_3 \cdot H_2O$ ,  $Na_2CO_3$ ,  $K_2CO_3 \cdot Na_2CO_3$ , and  
the isotherm of solv. at 150° consists of 4  
branches corresponding to cryst. of  $Na_2CO_3$ ,  $K_2CO_3 \cdot Na_2CO_3$ ,  
 $Na_2CO_3 \cdot H_2O$ , and  $K_2CO_3$ .  
Substance of  $K_2CO_3 \cdot Na_2CO_3$  at temper. is  
formed. Data show that  $K_2CO_3 \cdot Na_2CO_3$   
forms sq. veins, and at temp. below 150°,

$K_2CO_3 \cdot Na_2CO_3$  is  
over above 100°.  
Kaklova, Iosif.  
"Nauk. Khim."  
A study of solv.  
branches, corre-  
to 475° is con-  
tact be cryst.  
I. Marzina

(2) MZ

ITKINA, L.S.; KOKHOVA, V.F.

Solubility isotherms at 150° for systems:  $\text{Na}_2\text{SO}_4-\text{Na}_2\text{CO}_3-\text{H}_2\text{O}$  and  
 $\text{K}_2\text{SO}_4-\text{K}_2\text{CO}_3-\text{H}_2\text{O}$ . Izv. Sekt. fiz.-khim. anal. 26:242-247 1955.  
(MIRA 8:9)

1. Institut obshchey i neorganicheskoy khimii im. N.S. Kurnakova AN  
SSSR. (Sodium salts) (Potassium salts) (Solubility)

ITKINA, L.S.; KOKHOVA, V.P.

Solubility isotherm of the system  $2\text{Na}^+$ ,  $2\text{K}^+$  ||  $\text{CO}_3^{2-}$ ,  $\text{SO}_4^{2-}$  +  $\text{H}_2\text{O}$   
at  $150^\circ$ . Zhur.neorg.khim. 1 no.7:1665-1671 J1 '56. (MLRA 9:11)

(Sodium salts) (Potassium salts)

Kokhova, V.F.

USSR/Thermodynamics. Thermochemistry. Equilibria. Physico-Chemical B-8  
Analysis. Phase Transitions

Abs Jour : Ref Zhur - Khimiya, No 8, 1957, 26155

Author : L.S. Itkina, V.F. Kokhova

Inst : Academy of Sciences of USSR

Title : Solubility and Composition of Solid Phases in  $\text{Na}_2\text{SO}_4$  -  
 $\text{K}_2\text{SO}_4$  -  $\text{H}_2\text{O}$  System.

Orig Pub : Izv. Sektora fiz.-khim. analiza IONKh AN SSSR, 1956, 27,  
337-343

Abstract : The isotherm of solubility of the system  $\text{Na}_2\text{SO}_4$  (I) -  
 $\text{K}_2\text{SO}_4$  (II) -  $\text{H}_2\text{O}$  at  $150^\circ$  was studied by the method de-  
scribed earlier (RZhKhim, 1956, 18763; 42620). Three  
branches were revealed, which answer the crystallization of  
I, II and solid solutions on the basis of glaserite  $3\text{K}_2\text{SO}_4 \cdot$   
 $\text{Na}_2\text{SO}_4$  (III). The lattice of III is conserved in the solid  
solution at molar ratios of II: I of from 2.0 to 3.3, which

Card : 1/2

USSR/Thermodynamics. Thermochemistry. Equilibria. Physico-Chemical B-8  
Analysis. Phase Transitions

Abs Jour : Ref Zhur - Khimiya, No 8, 1957, 26155

has been confirmed by roentgenographic and crystallo-optic measurements. The compositions corresponding to joint crystallization at 150 and 120° were determined. The polytherm of the system from 0° to 150° was plotted. It shows a shift of the crystallization field of III into the region of solutions rich of II with the temperature rise.

Card : 2/2

28 (4)

AUTHORS:

Petrov, Ye. M., Kokhova, V. F.

05843

SOV/76-33-10-41/45

TITLE:

Small Autoclave With Thermostat and Mixer

PERIODICAL:

Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 10, pp 2372 - 2374  
(USSR)

ABSTRACT:

The authors designed a device consisting of a small autoclave, a thermostat, and a mixer used to stir the content of the autoclave (Fig 1). The device is used for investigations of phase equilibria of systems with liquid components at temperatures of above 100°C. The description indicates among other things that the autoclave has a removable reaction vessel with an additional container for the filtrate, and that filtration, i.e. separation of the liquid phase is brought about by centrifugation. All parts of the autoclave are made of stainless steel of the brands 1X189TM or 1X18N11B. The thermostat (Fig 2) is also described, which shows among other things that a contact thermometer with a 6P6-tube and an electron relay of the MKU-46RU 450 115 OD type is used for temperature control. The mixer is driven by a 2ASM-50 induction motor and consists of a holder for the autoclave and a reduction gear which is coupled with the electric motor. There are 2 figures.

Card 1/2

Small Autoclave With Thermostat and Mixer

05843  
SOV/76-33-10-41/45

ASSOCIATION: Nauchno-issledovatel'skiy institut teploenergeticheskogo priborostroyeniya (Scientific Research Institute for the Manufacture of Thermolectric Apparatus). Akademiya nauk SSSR, Institut organicheskoy i neorganicheskoy khimii, Moskva (Academy of Sciences of the USSR, Institute of Organic and Inorganic Chemistry, Moscow)

SUBMITTED: February 10, 1959

Card 2/2

ITKINA, L.S.; KOKHOVA, V.P.

Solubility and composition of the solid phases in the system  
 $\text{Na}_2\text{SO}_4 - \text{Na}_2\text{CO}_3 - \text{NaOH} - \text{H}_2\text{O}$  at 25 to 150°.. Zhur.neorg.khim. 5  
no.6:1290-1298 Je '60. (MIRA 13:7)  
(Systems (Chemistry))

ITKINA, L.S.; KOKHOVA, V.P.

Solubility isotherm for 150° in the system  $\text{Na}_2\text{SO}_4$  -  $\text{NaCl}$  -  $\text{NaOH}$  -  $\text{H}_2\text{O}$ .  
Zhur.neorg.khim. 5 no.9:2102-2110 S '60. {MIRA 13:11}  
(Sodium sulfate) (salt) (Sodium hydroxide)

LUK'YANOVA, Ye.I. [deceased]; KOKHOVA, V.F.

Solubility isotherm in the system NaOH - NaBH<sub>4</sub> - NaCl - H<sub>2</sub>O  
at 25°0. Zhur.neorg.khim. 8 no.1:218-225 Ja 4'63. (MIRA 16:5)  
(Sodium borohydride) (Salt) (Solubility)

GOLUBEV, T.M., doktor tekhn.nauk; DYADECHKO, G.P., inzh.; KOKHRYAKOV,  
B.D., kand.tekhn.nauk [deceased]

Investigating the process of vibration drawing. Met. i gornorud.  
prom. no.3:70-73 My-Je '62. (MIRA 15:9)

1. Khartsyzskiy staleprovolochno-kanatnyy zavod.  
(Wire drawing) (Vibrators)

KOKHRYAKOV, M. K.

"On fungal species and speciation."

report submitted for 10th Intl Botanical Cong, Edinburgh, 3-12 Aug 64.

All-Union Res Inst of Plant Protection, Leningrad.

KOKHTEV, A. A.

Technology

Standard river tugboats and barges: Standartigiz 1948.

9. Monthly List of Russian Accessions, Library of Congress, May 1957, Unc1.  
2

SENINA, R.M.; YURKOVA, N.I.; KOIKHTEV, A.A., inzhener, redaktor; BOBROVA, Ye.N., tekhnicheskiy redaktor.

[High-precision casting of measuring instrument parts; experience of the "Kalibr" plant] Vysokotchnoe lit's detalei ismeritel'nogo instrumenta; opyt zavoda "Kalibr." Moskva, Gos. nauchno-tekhn. izd-vo mashinostroit. lit-ry, 1951, 41 p.  
(Measuring instruments) (Die casting)

VESELOVSKIY, S.I.; KOKHTEV, A.A., redaktor; SHCHERBAKOV, P.V., tekhnicheskiy  
redaktor.

[Efficient process of manufacturing metal slitting and cutting saws]  
Ratsional'naya tekhnologiya izgotovleniya proresnykh i otresnykh  
diskovykh frez. Moskva, Gos.izd-vo obor.promyshl, 1953.25 p.[Microfilm]  
(Milling machines)

"APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000723710002-9

NEVZHRIN, V.K.; KOKHTEV, A.A., redaktor; ZUDAKIN, I.M., tekhnicheskiy re-  
daktor.

[Electric spark metal cutting] Elektroiskrovaya rasreshka metallov.  
Moskva, Gos. izd-vo oboronnoi promyshl., 1954. 98 p. (MIRA 8:2)  
(Metal cutting)

APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000723710002-9"

KOKHTEV, A. A. and STRAKHOV, A. P.

"Liquid-Cargo Barges from Interchangeable Standard Sections," Vest Mash, No 12,  
pp 21-31, 1954

Translation M-470, 18 May 55

USSR/Miscellaneous - Norms and standards

Card : 1/1 Pub. 128 - 31/32

Author : Koktev, A. A.

Title : Concerning the Leningrad conference on standardization

Periodical : Vest. mash. 34/7, 93 - 98, July 1954

Abstract : An industrial and technical conference was held in Leningrad in 1954, concerning the standardization, normalization, and unification of the Machine Construction Industry. Approximately 650 technical personnel from various branches of the Machine Construction Industry, and the Central Bureau of Standards, attended the conference. Subjects, which were discussed at the assembly, are presented.

Institution : ...

Submitted : ...

Kokhtev, A. A.

USSR/ Engineering - Barges

Card 1/1 Pub. 128 - 7/34

Authors : Strakhov, A. P., and Kokhtev, A. A.

Title : Cargo barges made of unified interchangeable sections

Periodical : Vest. mash. 12, 21-31, Dec 1954

Abstract : The editorial gives some information concerning the construction of river cargo-barges with load capacity of from 100 to 5,000 tons, and from 100 to 12,000 tons. Illustrations; diagrams; tables.

Institution : .....

Submitted : .....

KOKHTEV, A.A.

USSR/ Scientific Organisation - Conference

Card 1/1 Pub. 128 - 22/28

Author : Kokhtev, A. A.; Bug.

Title : Conference of readers "Herald of Machine Construction"

Periodical : Vest. nauch. 35/6, 83 - 85, Jun 1955

Abstract : A scientific conference was held in Feb. 1955, in Kiev, organized by the Regional Department of the All-Union Mechanical Engineers' Scientific and Technical Society, on problems connected with evaluation of technical and scientific literature published in the journal, "Herald for Machine Construction." Representatives of various machine construction plants, planning departments, scientific-research institutes, and the technical department of the Academy of Sciences of the Ukr. SSR, participated in the above mentioned conference. Abstracts of some lectures presented at the conference, are given.

Institution : ....

Submitted : ....

KOKHTEV, A.A.

MOSHIN, Ye.N., kandidat tekhnicheskikh nauk, redaktor; KOKHTEV, A.A., re-daktor; ZUDAKIN, I.M., tekhnicheskiy redaktor.

[Hammers without anvil blocks] Dostizheniya moloty. Moskva, Gos. izd-vo oboronnoi promyshlennosti, 1955. 79 p. (MIRA 8:6)  
(Hammers)